

Idaho State Police
Forensic Services
Toxicology Discipline



Section Four

Analysis of Alcohol and Common Volatile Solvents

4.1 Quantitative Analysis for Ethanol and Qualitative Analysis for Other Volatiles in Blood, Vitreous Humor and Urine by Dual Column Headspace Gas Chromatography

4.1.1 BACKGROUND

Fermented beverages such as beer and wine have been known and used by humans since prehistoric times.⁶ Ethanol abuse is often manifest in driving under the influence (DUI), a problem that plagues every country. The National Highway Traffic Safety Administration (NHTSA) estimates that alcohol was involved in 41% of fatal automobile crashes and 7% of all crashes in 1995.² Chronic alcoholism also contributes to ethanol related deaths. Ethanol consumed on a regular basis can lead to the development of alcoholic hepatitis which can progress into cirrhosis, liver failure, and death.^{2,6,7} Chronic excessive ingestion of ethanol is directly associated with serious neurologic and mental disorders such as brain damage, memory loss, sleep disturbances and psychoses.⁷ Alcohol is also involved in a high percentage of domestic disputes many of which result in injury and/or death.

Notwithstanding the public perception that ethanol is stimulatory, ethanol is classified as a *Central Nervous System Depressant*. Ethanol is a psychoactive drug that is similar in most respects to sedative-hypnotic compounds.⁴ The first mental processes to be affected are those that depend on training and previous experience.⁷ The individual's memory, concentration, and insight are dulled and subsequently lost. The person may become overly confident and exhibit uncontrolled mood swings and/or emotional outbursts.⁷ The effects of ethanol and other central nervous system depressants are additive, resulting in more sedation and greater impairment of driving ability.⁴

Ethanol is rapidly and completely absorbed from the stomach, small intestine and colon. The mechanism of absorption is a simple diffusion process, that is, alcohol moves from a region of higher concentration to a region of lower concentration.^{2,4,6} Alcohol is soluble in both water and fat, a property that facilitates its diffusion through biological membranes.⁴ The major amount of absorption takes place in the small intestine due to its large surface area, good blood supply and thin walled membrane. The time from the last drink to peak concentrations can range between 30 and 90 minutes, depending upon the individual's stomach contents.^{4,7} Alcohol absorption is slowed by the presence of food in the stomach. The time period required for gastric emptying is a prime factor that contributes to the wide variety of absorption rates of ingested ethanol observed in different individuals and under different conditions.^{2,7} Hence, the

extent of absorption in the stomach and small intestine is a function of the amount of ethanol at that site, the vascularity of the site and the surface area in contact with the blood supply.² Other factors that affect the absorption of ethanol include the type of beverage, the alcohol content, the rate of consumption and any disease state that affects normal gastric function or blood flow.²

Upon absorption, ethanol is distributed to all the water containing regions of the body. Within the blood there can be significant differences between arterial and venous blood depending upon the absorption status of the individual.² In the absorptive phase, the arterial blood ethanol concentration exceeds the venous blood ethanol concentration. Analysis of venous blood therefore, underestimates the brain alcohol concentration of the individual at this point. When absorption is complete there is little difference in ethanol concentration between arterial and venous blood.²

90 to 98 percent of ethanol is completely oxidized in the liver by reacting with the cofactor nicotinamide adenine dinucleotide (NAD) facilitated by alcohol dehydrogenase to produce acetaldehyde. Acetaldehyde is then acted upon by aldehyde dehydrogenase to form acetic acid which goes on to form carbon dioxide and water (figure 1). The amount of ethanol oxidized per unit time is roughly proportional to body weight and probably to liver weight. The remaining (unoxidized) alcohol is excreted unchanged in urine, expired air, saliva and sweat. The average elimination rate of ethanol is 0.015 g/dL/hour from men and 0.018 g/dL/hour for women.² In addition to gender, chronic abuse, prescription drugs and certain genetic factors can also influence the elimination rate.^{2,6,7}

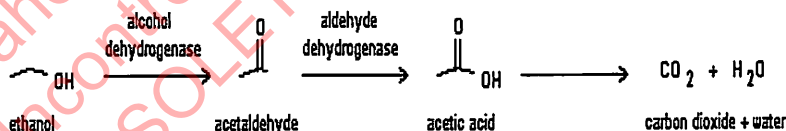


Figure 1. Metabolism of Ethanol.

Other commonly encountered alcohols such as methanol and isopropanol produce central nervous system (CNS) depressant effects but vary significantly in the degree. Methanol (wood alcohol) causes relatively little intoxication compared to ethanol.^{2,6} Its harmful effects are due to the direct result of its metabolism to formaldehyde (embalming fluid) and subsequently to formic acid. These metabolites lead to the destruction of neural cells, particularly the optic nerve, which can result in blindness.^{2,6} The formic acid leads to metabolic acidosis. Isopropanol (rubbing alcohol) is more toxic than ethanol with more prominent gastritis that includes pain, nausea, vomiting and hemorrhage.¹¹ Isopropanol is metabolized to acetone.

Toluene and acetone are commonly encountered in subjects that intentionally inhale ("huff") products including paint and contact adhesives to achieve effects akin to those of ethanol. The principal metabolite of toluene is benzoic acid.

Chronic abuse of toluene and/or acetone can lead to organ and CNS problems that may result in permanent damage.¹² Elevated endogenous acetone may be detected in the blood and urine of a diabetic or fasting individual. Acetone is metabolized to acetate and formate.

The analysis of ethanol and other volatiles in blood, vitreous humor and urine is accomplished with a gas chromatograph (GC) which uses a headspace analyzer for sample introduction. Samples, controls and standards are sealed into vials that contain an aqueous n-propanol internal standard solution and heated by the headspace analyzer. As described in Henry's Law, in a closed container at a given temperature, a direct (proportional) relationship exists between the amount of a volatile substance dissolved in a liquid and the amount of the volatile substance in the headspace vapor above the solution. An aliquot of the vapor is injected into the GC in a dual column configuration. The GC serves to separate out the components of the solution as a function of their chemical properties. Separated components are detected by a flame ionization detector (FID). The qualitative identification of ethanol and other common volatiles is based on the retention time determined, relative to the n-propanol internal standard, for each of the columns. This method also provides for a quantitative determination for ethanol. The quantitative result is based on a minimum of a three-point calibration curve, which uses the peak area ratio between ethanol and the n-propanol internal standard.

4.1.2 SCOPE

This method describes the Idaho State Police Forensic Services (ISP-FS) standard operating procedure for the analysis of aqueous samples (blood, vitreous humor and urine) for the presence of volatile compounds including methanol, ethanol, acetaldehyde, acetone, isopropanol, toluene and related compounds, via a headspace sampling gas chromatographic method. The temperature program for Gas Chromatograph must provide for baseline separation of volatile compounds of interest as indicated by analysis of multicomponent mixtures. IDAPA 11.03.01 requires that determinations for ethyl alcohol for legal purposes be performed on blood samples.⁸ For this reason, serum is not an appropriate sample under IDAPA 11.03.01. If serum is analyzed a qualifier advising that the sample does not comply with IDAPA 11.03.01 must be placed on the analysis report.

4.1.3 EQUIPMENT

4.1.3.1 Perkin Elmer Auto System XL Gas Chromatograph (GC)

4.1.3.2 Columns

4.1.3.2.1 Restek Rtx[®]-BAC1 (#18003: 30 meter X 0.32mm inner diameter (ID), 1.8µm film thickness (FT) or equivalent column)

4.1.3.2.2 Restek Rtx[®]-BAC2 (#18002: 30 meter X 0.32mm ID, 1.2 µm FT or equivalent column)

4.1.3.3 Perkin Elmer HS-40 or HS-110 Headspace Autosampler (figures 2 and 3)

- 4.1.3.4 PE Workstation Software, TotalChrom Version 6.2.0 or more recent version/upgrade.
- 4.1.3.5 Headspace (HS) vials (P-E B0104236 or equivalent)
- 4.1.3.6 Safety Closures for HS vials {PTBE coated rubber septa, crimp caps and star springs} (P-E B0104240 or equivalent)



Figure 2. HS-40



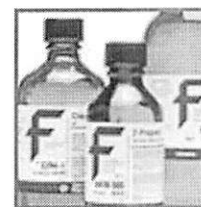
Figure 3. HS-110

- 4.1.3.7 Hand Crimper or Bench Top Crimper
- 4.1.3.8 Hamilton MICROLAB 503A or equivalent semi-automatic Dilutor/Pipetter equipped with sample and reagent syringes capable of dispensing 250 μ L and 2000 μ L, respectively.

4.1.4 REAGENTS

A certificate of analysis should be requested for all chemicals.

- 4.1.4.1 1-Propanol (Acros/Fisher Scientific # 23207-0010, #A996-1 or equivalent)
- 4.1.4.2 Acetone (Fisher #A18-500 or equivalent)
- 4.1.4.3 Acetaldehyde (Fisher #01004-250 or equivalent)
- 4.1.4.4 Isopropanol (2-Propanol) (Fisher #A416-500 or equivalent)
- 4.1.4.5 Ethanol, 200 proof (Sigma 45,984-4)
- 4.1.4.6 Methanol (Fisher #A412-500 or equivalent)
- 4.1.4.7 Toluene (Fisher T324-500 or equivalent)
- 4.1.4.8 Ammonium Sulfate (Fisher #A702-500 or equivalent)
- 4.1.4.9 Sodium Fluoride (Fisher #S299-500 or equivalent)



4.1.5 REFERENCE MATERIAL

4.1.5.1 Calibrators

Ethanol reference material, with nominal values from 0.01 to 0.50g/100cc, to establish the calibration curve can be obtained through Cerilliant, EM Science, NIST or other reputable vendor. The requirements for calibrators are discussed in 4.1.7.2.

- 4.1.5.2 Internal Standard Solution - 0.03g/dL 1-propanol in 1.0M (NH₄)₂SO₄
 Add approximately 800mL of DI water to a 1L volumetric flask. Add 132.14g (NH₄)₂SO₄ and mix to dissolve.
 Add 1g sodium fluoride {optional}.
 Add 375µL 1-propanol.
 QS to 1L with distilled water.
 Record preparation on reagent log.

Solution is stable for 2 months.

4.1.5.3 Qualitative Volatile Standard Multicomponent Mix Solution

4.1.5.3.1 Option One

Add approximately 200mL of DI water to a 250mL volumetric flask.

Add one or more of the following volatiles, *as needed*:

100 µL acetaldehyde

100 µL acetone

500 µL methanol

500 µL isopropanol

500 µL ethanol

50µL toluene

QS to 250mL.

Record preparation on reagent log.

Solution is stable for 1 year when stored under refrigeration.

4.1.5.3.2 Option Two

Obtain Cerilliant #A-054 or comparable.

Solution contains acetone, ethanol, methanol and isopropanol standards, which this method utilizes qualitatively. Ethanol levels are at 0.05, 0.10 or 0.40g/100cc. This solution may simultaneously serve as an aqueous ethanol control.

4.1.5.4 Controls

4.1.5.4.1 Whole Blood Volatiles Control (g/100cc)

BIO-RAD 565 or equivalent.

4.1.5.4.1.1 Level 1

≅0.08 ethanol.



4.1.5.4.1.2 Level 2

≅0.15 ethanol.

Level 2 also contains acetone, methanol and isopropanol standards, which this method utilizes qualitatively.

4.1.5.4.2 **Aqueous Control (g/100cc)**

4.1.5.4.2.1 Ethanol

Ampules containing ethanol reference material, with nominal values from 0.01 to 0.50g/100cc, for use as a control can be obtained through Cerilliant, EM Science, NIST or other reputable vendor. Ideally the source should differ from the one used to prepare calibrators.

4.1.6 **SAFETY CONCERNS**

Blood samples should be processed and chemicals handled according to safety guidelines in the *Idaho State Police Forensic Services Health and Safety Manual*.

4.1.7 **QUALITY ASSURANCE**

4.1.7.1 General

4.1.7.1.1 While at the laboratory blood or vitreous samples are to be stored under refrigeration and urine samples can be either refrigerated or frozen. Urine samples should be frozen for long-term storage.

4.1.7.1.2 The syringes on the Pipetter/Dilutor must be checked for accuracy and precision. Refer to toxicology manual section 5.1 for pipette calibration requirements and options.

4.1.7.1.3 Refer to toxicology manual section 5.2 for balance calibration requirements.

4.1.7.1.4 Refer to toxicology manual section 5.3.2 for GC-HS maintenance schedule.

4.1.7.1.5 Current source and lot number of controls and reference material must be maintained on *Volatiles Analysis Calibration Run and Volatiles Analysis Non-calibration Run* spreadsheet forms.

4.1.7.2 Calibration Requirements

4.1.7.2.1 A minimum of three ethanol calibrators must be used to establish a calibration curve.

- 4.1.7.2.2 The calibrators chosen must characterize the entire range of interest.
- 4.1.7.2.3 Ethanol calibrators should be analyzed in order of increasing concentration.
- 4.1.7.2.4 The least squares line resulting from the analysis of the ethanol calibrators must have a coefficient of correlation of ≥ 0.998 .
- 4.1.7.2.5 Each ethanol calibrator may have more than one replicate.
- 4.1.7.2.6 When a calibration standards are run in duplicate, if one of the replicates is not usable, a single replicate can be used to establish the response factor.
- 4.1.7.2.7 A calibration is valid for two weeks provided all values for controls fall within acceptable ranges.
- 4.1.7.2.8 Once established, analysts not involved in establishing the calibration curve may use the calibration.
- 4.1.7.2.9 An analysis run may include samples prepared by more than one analyst.

4.1.7.3 Per Analysis Run Control Requirements

4.1.7.3.1 **Calibration Run**

- 4.1.7.3.1.1 An internal standard blank should follow the last ethanol calibrator.
- 4.1.7.3.1.2 A water blank may be included in each calibration run.
- 4.1.7.3.1.3 For up to 10 samples (20 vials), an analysis run must include either a high or low blood control in duplicate before proceeding with additional samples.
- 4.1.7.3.1.4 For analysis run consisting of more than 10 samples (20 vials), a minimum of one blood or aqueous control must be run with each additional 10 samples (20 vials).

- 4.1.7.3.1.5 The blood control concentration(s) should be varied periodically.
- 4.1.7.3.1.6 A blood or aqueous control containing ethanol, with or without other volatile substances, meets the "per run" requirement.
- 4.1.7.3.1.7 Each calibration run should include the Volatile Standard Mix Solution (4.1.5.3.1) or the Multicomponent Alcohol Calibration Kit solution (Cerilliant A-054 or equivalent).
- 4.1.7.3.1.8 A commercially obtained multicomponent volatile mix may be used as both an aqueous ethanol control and a multicomponent mixture.
- 4.1.7.3.2 **Non-Calibration Run**
- 4.1.7.3.2.1 The same batch of internal standard used in the calibration run must be used for additional runs within the two-week period.
- 4.1.7.3.2.2 An internal standard blank should follow a the analysis of the high blood control.
- 4.1.7.3.2.3 The analysis run must include both high and low blood controls in duplicate.
- 4.1.7.3.2.4 When the analysis exceeds 10 samples (20 vials), a minimum of one blood or aqueous control must be run with each additional 10 samples (20 vials).
- 4.1.7.3.2.5 A blood or aqueous control containing ethanol, with or without other volatile substances, meets the "per run" requirement.
- 4.1.7.3.2.6 Additional aqueous controls can be included at the discretion of the analyst.

4.1.7.3.3 Monitoring of Control Values

On an ongoing basis, calculate the mean and standard deviation of control samples. The data serve as a continual check of target values.

4.1.8 ANALYSIS PROCEDURE

4.1.8.1 General

4.1.8.1.1 Bring calibrators, controls, internal standard and samples to room temperature.

4.1.8.1.2 Gather necessary vials, closures and ancillary supplies in or near laminar flow hood or biological safety cabinet.

4.1.8.1.3 Sample preparation should take place in a laminar flow hood.

4.1.8.2 Collection Kit Description and Labeling

4.1.8.2.1 Note sample container description on *Volatiles Analysis Coversheet*.

4.1.8.2.2 Inspect inner seals on specimen container and note their condition on *Volatiles Analysis Coversheet*.

4.1.8.2.3 Note condition of seals and initial on toxicology submittal form if version of form requires analyst input.

4.1.8.2.4 Laboratory number should be placed on each sample container.

4.1.8.2.5 When more than one sample is present, label samples "A", "B", etc., or comparable. Note container type(s) on analysis coversheet.

4.1.8.3 Case Sample and Collection Container Evaluation

4.1.8.3.1 Note sample type (blood, urine, or vitreous humor) on *Volatiles Analysis Coversheet*.

4.1.8.3.2 When the sample is blood or urine, the sample container (tube or bottle) and contents should be examined to determine compliance with Idaho Administrative Code, IDAPA 11.03.01.⁸

4.1.8.3.3 IDAPA 11.03.01 does not apply to specimens collected from deceased individuals, however, a qualifier statement, advising how the integrity of the sample is

adversely affected, may be added to the report as described in section 4.1.8.13.2.

4.1.8.3.4 **Blood Specimen Container Evaluation**

4.1.8.3.4.1 IDAPA requires that blood specimens be collected in a container, which contains ten (10) milligrams of sodium fluoride per cubic centimeter of blood plus an appropriate anticoagulant.

4.1.8.3.4.2 The containers provided in ISP-FS kits comply with these requirements. It should, however, not be assumed that an ISP-FS kit contains the container it was supplied with.

4.1.8.3.4.3 Non-ISP-containers should be evaluated as to compliance. Information from the manufacturer of the container will indicate sodium fluoride concentration as well as the presence of an appropriate anticoagulant.

4.1.8.3.4.4 Note specimen container compliance on *Volatiles Analysis Coversheet*.

4.1.8.3.5 **Urine Specimen Container Evaluation**

4.1.8.3.5.1 IDAPA requires that urine specimens be collected in clean, dry containers.

4.1.8.3.5.2 The containers provided in ISP-FS kits comply with these requirements. It should, however, not be assumed that an ISP-FS kit contains the container it was supplied with.

4.1.8.3.5.3 Note type of specimen container on *Volatiles Analysis Coversheet*.

4.1.8.3.6 **Blood Specimen Evaluation**

4.1.8.3.6.1 IDAPA 11.03.01 describes blood to be whole blood. When it is the analyst's opinion that the intended blood sample is serum or otherwise questionable, or with a questionable urine sample, the analyst has the three options.

- 4.1.8.3.6.2 Option One
The sample is not analyzed. A comment "Sample unsuitable for analysis" is placed on the analysis report.
- 4.1.8.3.6.3 Option Two
The sample is analyzed in accordance with this SOP. A qualifier is placed on the analysis report, "The sample does not comply with IDAPA 11.03.01".
- 4.1.8.3.6.4 Option Three
The sample is analyzed for ethanol and other volatiles according to SOP 4.2, Analysis of Solutions Containing Ethanol and Common Volatiles. The report, therefore, will make no mention of the sample having a biological origin.
- 4.1.8.4 Pipetter/Dilutor Set-up
- 4.1.8.4.1 Switch on power.
- 4.1.8.4.2 Display will inquire as to the sizes of installed syringes. Select the correct size for sample syringe [right] and reagent syringe [left].
- 4.1.8.4.3 Scroll down to volume option. Select 250 μ L for sample syringe [right] and 2000 μ L for reagent syringe [left].
- 4.1.8.4.4 Scroll down to speed option. Verify that syringe speed is on desired setting.
- 4.1.8.4.5 Prime the fluid path. Continue priming until no bubbles are observed.
- 4.1.8.5 Preparation for Case Sample Analysis
- 4.1.8.5.1 Label two HS vials with the laboratory number.
- 4.1.8.5.2 Place sample container on rocker for a minimum of two minutes.
- 4.1.8.5.3 If the blood sample appears to be coagulated, the sample may require homogenization in a tissue grinder, or equivalent.

4.1.8.5.4 Use Pipetter/Dilutor dispense 250 μ L of sample and 2000 μ L of internal standard (ISTD) to a labeled headspace vial.

4.1.8.5.5 Seal headspace vials **immediately** with crimp caps as illustrated in Figure 4. The gray side of the septum should be facing down.

4.1.8.6 Preparation of Calibration Standards

4.1.8.6.1 Label HS vials for with each ethanol calibration standard.

4.1.8.6.2 Use Pipetter/Dilutor to dispense 250 μ L of aqueous ethanol standard and 2000 μ L of internal standard (ISTD) into each labeled headspace vial.

4.1.8.6.3 Seal **immediately** with crimp cap as illustrated in Figure 4 (or equivalent).

4.1.8.7 Preparation of Quality Assurance Samples

4.1.8.7.1 Water Blank

4.1.8.7.1.1 Label HS vial for *water blank*.

4.1.8.7.1.2 Add \cong 2000 μ L DI water to labeled headspace vial.

4.1.8.7.1.3 Seal **immediately** with crimp cap as illustrated in Figure 4 (or equivalent).

4.1.8.7.2 Internal Standard Blank

4.1.8.7.2.1 Label HS vial for *ISTD blank*.

4.1.8.7.2.2 Use Pipetter/Dilutor to dispense 2000 μ L of internal standard (ISTD) into labeled headspace vial.

4.1.8.7.2.3 Seal **immediately** with crimp cap as illustrated in Figure 4 (or equivalent).

4.1.8.7.3 Blood and Aqueous Controls

4.1.8.7.3.1 Label HS vials for appropriate whole *blood and aqueous controls*.

4.1.8.7.3.2 Use Pipetter/Dilutor to dispense 250 μ L control and 2000 μ L internal standard (ISTD) into each labeled headspace vial.

- 4.1.8.7.3.3 Seal **immediately** with crimp cap as illustrated in figure 4 (or equivalent).

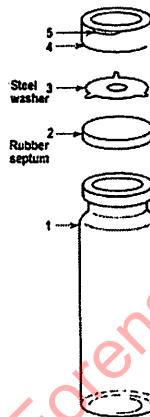


Figure 4. Crimp cap assembly

4.1.8.8 Preparation for Run

4.1.8.8.1 **Open Sequence Editor.**

4.1.8.8.2 Into Sequence log table, enter the sample case numbers, ethanol standards, other volatiles mix, blanks and controls.

4.1.8.8.3 Load samples, calibration standards, blank and controls into the carousel of the headspace sampler as noted in the sequence table.

4.1.8.9 Analysis Parameters

4.1.8.9.1 Refer to instrument METHOD printouts for gas chromatograph and headspace analyzer analysis parameters.

4.1.8.9.2 The method should be set up such that all samples (casework, calibrators, and controls) are quantitated to a minimum of three decimal places (0.000).

4.1.8.9.3 Analysis method printouts must be stored centrally.

4.1.8.10 Evaluation of Analysis Run

4.1.8.10.1 **Qualitative Accuracy Criteria**

The qualitative presence of ethanol or other volatile substance can be established if the relative retention time (RRT) for a specimen is within ± 0.10 minutes of the RRT of a standard of the compound in question. This rejection criterion should be designated in the data station analysis method.

4.1.8.10.2 Quantitative Accuracy Criteria

4.1.8.10.2.1 *On Volatiles Analysis Calibration Run or Volatiles Analysis Non-calibration Run* spreadsheet form(s), determine the mean ethanol concentration for each calibrator and control sample. Spreadsheet should be formatted such that ethanol values are calculated and truncated using a minimum of three decimal places.

4.1.8.10.2.2 The quantitative ethanol results for a batch of samples can be accepted if the values obtained for control samples fall within $\pm 10\%$ of target value. Target values are determined as described in section 4.1.10.2.

4.1.8.10.3 Quantitative Precision Criteria

4.1.8.10.3.1 Column Precision

On Volatiles Analysis Calibration Run or Volatiles Analysis Non-calibration Run spreadsheet form(s), determine the column precision for each case, calibrator or control sample. The values obtained from column 1 and column 2 should agree within 0.015g/100cc. If the precision requirement is not met, the sample must be reanalyzed.

4.1.8.10.3.2 Replicate Precision

When more than one replicate for a case, calibrator or control sample, is analyzed, calculate the mean value (columns 1 and 2) for each sample with volatiles analysis spreadsheet form. The mean value for replicates must agree as described in the following table. If the precision requirement is not met, the sample must be reanalyzed.

Results Range (g/100cc)	Precision (g/100cc)
0.02 - 0.10	0.01
0.11 - 0.20	0.015
0.21 - 0.40	0.020

4.1.8.10.3.3 For blood samples, homogenization should be considered when a lack of precision is observed.

4.1.8.10.4 High Ethanol Values

4.1.8.10.4.1 When an elevated ethanol value is obtained, appropriate calibrators must bracket the value. For instance, if a 0.38g/100cc blood ethanol value is determined, a 0.40g/100cc calibrator must be included in the run. The analyst has the option to either re-analyze the sample with a 0.40 calibrator or dilute the sample by 0.5.

4.1.8.10.4.2 All samples above 0.40g/100cc two options are available.

Option One

Sample may be reanalyzed with a 0.5 dilution. The dilution factor should be incorporated into final calculations.

Option Two

No further analysis with the report stating "greater than 0.40g/100cc".

4.1.8.11 Reporting of Quantitative Ethanol Results

4.1.8.11.1 **Blood**

4.1.8.11.1.1 Samples must be quantitated to a minimum of three decimal places (0.000).

4.1.8.11.1.2 Report truncated mean value, of grams of ethanol per 100cc of blood, to two decimal places (0.00).

4.1.8.11.1.3 Report values <0.02g/100cc as **none detected**.

4.1.8.11.1.4 If the sample and/or sample vial does not comply with IDAPA 11.03.01, an

appropriate comment should be noted on the analysis report. Examples of comments are described in section 4.1.8.13.

4.1.8.11.2 **Urine**

4.1.8.11.2.1 Samples must be quantitated to a minimum of three decimal places.

4.1.8.11.2.2 Result obtained is be multiplied by 0.67. Report truncated mean value, as grams of ethanol per 67 mL of urine, to two decimal places.

4.1.8.11.2.3 Report values $<0.02\text{g}/67\text{mL}$ as "*none detected*".

4.1.8.11.2.4 A qualifier statement "*Urine results may be of questionable value*" must be included in the report.⁸

4.1.8.11.3 **Vitreous Humor**

4.1.8.11.3.1 Samples must be quantitated to a minimum of three decimal places.

4.1.8.11.3.2 Report truncated mean value, as grams of ethanol per 100mL of vitreous humor, to two decimal places.

4.1.8.11.3.3 No conversion to a blood alcohol value should be made.

4.1.8.11.3.4 Report values $<0.02\text{g}/100\text{cc}$ as "*none detected*".

4.1.8.12 Reporting of Qualitative Volatiles Results

The qualitative presence of other volatiles such as acetone, isopropyl alcohol, methyl alcohol, toluene and formaldehyde should be noted on the analysis report following the ethyl alcohol results.

4.1.8.13 Comments for Analysis Report

As appropriate and/or required, comments outlining actions, discrepancies and/or qualifiers should be included on the analysis report following the results of analysis.

4.1.8.13.1 **Living Subjects or Unknown Status**

- *Collection tube(s) do not comply with IDAPA 11.03.01.*

This comment only applies when blood is collected since IDAPA does not address the collection tube for vitreous humor.

- *Sample does not comply with IDAPA 11.03.01.*

This comment should be used when the sample is clearly not blood.

4.1.8.13.2 Deceased Subjects

- *The lack of appropriate preservative in collection tube(s) makes this result questionable.*

- *The lack of appropriate preservative in collection tubes could adversely affect the reliability of this result.*

These comments can be applied to both blood and vitreous humor samples not collected in an IDAPA approved container. The IDAPA requirements only apply to evidentiary testing for living subjects; however, the high potential for ethanol production in non-preserved samples from deceased individuals makes a qualifier statement necessary.

4.1.8.13.3 General Comments

- *Collection kit forwarded for further analysis.*

Specify where the kit will be forwarded.

- *Specimen unsuitable for testing.*

This comment can be used with questionable samples as described in section 4.1.8.3.6.2.

4.1.9 ANALYSIS DOCUMENTATION

4.1.9.1 Volatiles Analysis Forms

- #### 4.1.9.1.1
- Required forms are located on the I:\ drive under TOXICOLOGY PROGRAM FORMS\Alcohol Analysis Forms\:

...\Volatiles Analysis Coversheet.doc

...\Volatiles Analysis Calibration Run.xls

...\Volatiles Analysis Non-calibration Run.xls

- #### 4.1.9.1.2
- The non-calibration run form is used for both run continuation (HS-40) and additional analysis runs within the two week time period.

- 4.1.9.1.3 The *Volatiles Analysis Calibration Run* form must be included in all casefiles.
- 4.1.9.1.4 The *Volatiles Analysis Calibration Run* and *Volatiles Analysis Non-calibration Run* forms should be updated whenever instrument parameters are adjusted.
- 4.1.9.1.5 The *Volatiles Analysis Calibration Run* and *Volatiles Analysis Non-calibration Run* forms formatting must be such that ethanol values are reported to a minimum of three decimal places.

4.1.9.2 Casefile Contents

Each analysis casefile must contain a *Volatiles Analysis Coversheet*, *Volatiles Analysis Calibration Run* and when appropriate a *Volatiles Analysis Non-calibration Run* spreadsheet form and the analysis data printouts for the particular case sample.

4.1.9.3 Quality Assurance Data

4.1.9.3.1 A copy of controls and standards need not be included in individual case files.

4.1.9.3.2 A packet containing data for response factor/calibration curve, controls and standards will be prepared for each analysis run and stored centrally in the file designated for alcohol quality assurance data in the laboratory where the analysis was performed until archiving.

4.1.9.3.3 When necessary, a copy of the control and standard printouts can be prepared from the centrally stored documents.

4.1.10 AUTHENTICATION OF REFERENCE MATERIALS

4.1.10.1 Quantitative and Qualitative Volatile Standards

4.1.10.1.1 Standards for quantitative purposes must be traceable to NIST standards (or comparable).

4.1.10.1.2 *Certificate of Analysis* for all reference standards will be stored centrally.

4.1.10.1.3 New lots of aqueous ethanol, aqueous mixed volatiles, and volatile reagent standards should be included in duplicate in a minimum of one analysis run prior to official use.

4.1.10.1.4 Standards authenticated prior to the start date of this SOP revision can be used until consumed. The authentication data must be centrally stored.

4.1.10.1.5 **Aqueous Ethanol Standards**

4.1.10.1.5.1 The *Certificate of Analysis*, together with a comparison of relative retention time and quantitation data, against existing calibrators, will serve as the qualitative and quantitative authentication of ethanol in the standard.

4.1.10.1.5.2 The new lot number can be accepted if the mean relative retention time for the new standard is ± 0.10 minutes and the mean concentration obtained falls within 6% of the target value listed on the *Certificate of Analysis*.

4.1.10.1.6 **Aqueous Mixed Volatile Standards**

4.1.10.1.6.1 The Certificate of Analysis for an aqueous mixed volatile standard along with a comparison to data from the previous runs will serve as the qualitative authentication of the components of the standard.

4.1.10.1.6.2 The solution prepared with a new lot number of volatile chemical standard can be accepted if the mean relative retention time for the new standard is ± 0.10 minutes. If ethanol is present in the mixture, the mean concentration must fall within 6% of the target value listed on the *Certificate of Analysis*.

4.1.10.1.7 **Volatile Reagent Standards**

4.1.10.1.7.1 For reagent standards (acetone, ethanol, methanol, isopropanol, toluene,...) used to prepare single constituent or mixed standard of volatiles, the qualitative authentication is established with the *Certificate of Analysis* and comparison of relative retention times.

4.1.10.1.7.2 The new lot number can be accepted if the mean relative retention time (RRT) for the

new standard is ± 0.10 minutes from the RRT of existing the qualitative standard components. The reagent should be diluted prior to analysis as described in section 4.1.7.2.

4.1.10.2 Blood Controls

4.1.10.2.1 The Toxicology Program Discipline Leader or designee will characterize a new lot prior to release to individual analysts.

4.1.10.2.1.1 The new lot will be analyzed in each ISP-FS laboratory involved in alcohol analysis.

4.1.10.2.1.2 Each ISP-FS laboratory will provide a minimum of 20 determinations.

4.1.10.2.1.3 The target value and range of a blood control lot will be established through establishing a mean of all provided determinations.

4.1.10.2.1.4 The new blood lot number can be accepted if the mean relative retention time for the new control is ± 0.10 minutes of the RRT currently established for ethanol and the mean concentration obtained falls within the range provided in the manufacturer's package insert.

4.1.10.2.1.5 A 10% range will be calculated from the mean value of the determinations and used to evaluate accuracy on subsequent analysis. 5% should serve as a warning limit.

4.1.10.2.1.6 For blood controls that contain other volatiles (acetone, methanol, isopropanol, toluene) in addition to ethanol; the qualitative determination of the components should be established through the comparison of relative retention times from the previous run. The values should agree within ± 0.10 minutes.

4.1.10.2.2 Blood control *Package Inserts* will be stored centrally.

4.1.11 REFERENCES

- 4.1.11.1 Stafford, D.T., *Chromatography. in: Principles of Forensic Toxicology*, edited by Barry Levin, pp. 93-101, 103-114, AACC Press, 1999.
- 4.1.11.2 Levine, B., *Alcohol. in: Principles of Forensic Toxicology*, edited by Barry Levin, pp. 170-184, AACC Press, 1999.
- 4.1.11.3 Caplan, Y.H., *The Determination of Alcohol in Blood and Breath. in: Forensic Science Handbook*, edited by Richard Saferstein, pp. 594-648, Prentice-Hall New Jersey, 1982.
- 4.1.11.4 Julien, R.M., *Central Nervous System Depressants: Alcohol and the Inhalants of Abuse. in: Primer of Drug Action*, pp. 64-92, Freeman-New York, 1998.
- 4.1.11.5 Saker, E.G., *Screening and Quantitation by Head Space Technique of Some of the Vapors Most Commonly Found in Forensic Toxicology*, in: *Current Approaches in Forensic Toxicology*, Chapter 11, SOFT Meeting, 1994.
- 4.1.11.6 Perrine, D.M., *Depressants: Alcohol, Benzodiazepines, Barbiturates. in: The Chemistry of Mind-Altering Drugs*, pp. 113-129, ACS, Washington, DC, 1996.
- 4.1.11.7 Hobbs, W.R., Rall, T.W. and Verdoorn, T.A., *Drugs Acting on the Central Nervous System - Hypnotics and Sedatives; Ethanol. in: Goodman and Gilman's The Pharmacological Basis of Therapeutics*, pp. 361, 386-393, McGraw-Hill, 1996.
- 4.1.11.8 Idaho Administration Code, IDAPA 11.03.01, Rules Governing Alcohol Testing.
- 4.1.11.9 Christmore, D.S., Kelly, R.C. and Doshier, L.A. *Improved Recovery and Stability of Ethanol in Automated Headspace Analysis*, *J. Forensic Sci.* 29(4): 1038-1044; 1984.
- 4.1.11.10 Restek Applications Note #59598, Dual-Column Confirmational GC Analysis of Blood Alcohols Using the Rtx[®]-BAC1 and Rtx[®]-BAC2 Columns Optimized for the Perkin-Elmer HS-40 Headspace Autosampler, 1999.
- 4.1.11.11 Klaassen, C.D., *Nonmetallic Environmental Toxicants. in: Goodman and Gilman's The Pharmacological Basis of Therapeutics*, pp. 1681-1682, McGraw-Hill, 1996.

- 4.1.11.12 Klaassen, C.D., *Inhalants. in: Principles of Forensic Toxicology*, edited by Barry Levin, pp. 341-348, AACC Press, 2003.
- 4.1.11.13 Baselt, R.C., *Ethanol. in: Disposition of Toxic Drugs and Chemicals in Man*, pp. 411-414, Biomedical Publications, 2004.

Property of Idaho State Police Forensic Services
Uncontrolled Internet Copy
OBSOLETE DOCUMENT

**Idaho State Police
Forensic Services
Toxicology Discipline**



Section Four
Analysis of Alcohol and Common Volatile Solvents

4.1 Quantitative Analysis for Ethanol and Qualitative Analysis for Other Volatiles in Blood, Vitreous Humor and Urine by Dual Column Headspace Gas Chromatography

Revision #	Issue Date	History
0	10/01	
1	05-15-02	Clarifications, coefficient of correlation change for system compatibility.
2	09-13-02	Addition of analysis documentation section.
3	01-03-03	Clarifications, refinement of analysis documentation section 4.1.10.
4	04-06-04	Clarifications, acceptance criteria and quality assurance sections amended, authentication guidelines added.
5	06-16-04	Addition to section 4.1.8.5.3. Modification of 4.1.8.11.4 (<i>duplicate</i> replaced with <i>replicate</i>)
6	12-29-2005	Modified format, updated and clarified quality assurance requirements.

Approval

Discipline Leader: _____ **Date:** _____
Susan C. Williamson

Issuance

QA Manager: _____ **Date:** _____
Richard D. Groff